## Supplementary Material

for

Deaminatively-Generated Carbocations as Initiators of Styrene Polymerization.

by

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Materials and methods. Reagents were purchased from the Aldrich Chemical Company; most were used without further purification. Styrene was vacuum distilled prior to use. Hexane was distilled before using.

NMR spectra were recorded on a Bruker AMX 300 MHz, FT instrument; UV-Vis and IR spectra were measured using a Beckman Model 25 UV-Vis Spectrometer and a Perkin Elmer 1600 Series FT-IR spectrometer, respectfully. All syntheses and reactions of the N-nitrosoamide were performed in the dark. The Parr pressure reactor (equipped with a stirrer and temperature regulator) was purged with and then filled with argon prior to use. Clean, vacuum-dried syringe/needle assemblies were flushed 10 times with argon before using.

Stability of the Precursor; Handling and Storage. N-Nitrosopivalamides in addition to being thermolabile are labile in the presence of acids, bases and moisture. As a result, the nitrosoamide was freshly prepared for each run. All manipulations involving the nitrosoamide were performed in the dark. Caution! Nitrosoamides should be handled with extreme care because of their possible mutagenicity <sup>1a</sup> and carcinogenicity (local and

systemic). <sup>1b</sup> Efficient fume hoods and appropriate personal protection (chemical-resistant gloves, safety glasses, lab coat, etc.) are recommended when handling these compounds.

N-4-Methoxybenzylpivalamide was prepared from the procedure of Heyns and von Bebenburg. <sup>2</sup>: m.p. 81°-82°C (lit <sup>3</sup> 81°-82°C); IR (KBr) 3309, 1689 1510, 1390, 1375 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.27 (s, 9H), 4.44 (d, 2H, J = 7Hz), 5.90 (bs, 1H), 7.26-7.32 (m, 5H),  $\lambda_{max}$  = 284 ( $\epsilon$  = 209).

N-(4-Methoxy)-benzyl-N-nitrosopivalamide (1). A mixture of N-4-methoxybenzyl pivalamide (955 mg, 5 mmol), NaOAc (2.5 g, 30 mmol), and Na<sub>2</sub>SO<sub>4</sub> (5 g) was dried at oil pump vacuum. Methylene chloride (30 cm<sup>3</sup>) freshly distilled from  $P_2O_5$ , was added, under  $N_2$ , to the solid material and the suspension was cooled to  $-78^{\circ}$ C A solution of  $N_2O_{4(l)}$  (2 cm<sup>3</sup>, 31 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 cm<sup>3</sup>) at -78 °C was then added to the stirred suspension at -78°C which was then allowed to warm to -25°C over 20 minutes. After a further 20 minutes at -25°C, the suspension was washed rapidly in turn with saturated solutions of NaCl, NaHCO<sub>3</sub> and NaCl at -5°C. The organic phase was dried with stirring over Na<sub>2</sub>SO<sub>4</sub> at -30 °C and was evaporated *in vacuo* for ~20 minutes at -30°C to yield a lemon yellow oil (1.1 g, 5 mmol, 100%): IR (Neat) 1720, 16059, 1502, 1390, 1375 cm<sup>-1</sup>; <sup>1</sup>H NMR (CD<sub>3</sub>CN)  $\delta$  1.45 (s, 9H), 4.97 (s, 2H), 7.05-7.40 (m, 5H). UV (CH<sub>2</sub>Cl<sub>2</sub>)  $\lambda_{max}$  275 nm ( $\epsilon$  = 500), 400 nm ( $\epsilon$  = 63), 394 nm (sh), 422 nm ( $\epsilon$  = 66).

Decomposition of N-(4-methoxy)-benzyl-N-nitrosopivalamide (1) in Styrene. In a typical run, cyclohexane (100 cm<sup>3</sup>) and styrene (10 cm<sup>3</sup>, 87.2 mmol) were introduced via argon-purged syringes into a 500 cm<sup>3</sup> argon-purged Parr reactor. N-(4-Methoxy)-benzyl-N-nitrosopivalamide (1) (0.25 g, 1.2 mmol) in cyclohexane (10 cm<sup>3</sup>) was then

introduced via syringe into the reactor; the solution was stirred vigorously and the temperature was maintained at 25°C for 3h. After 3h, the reaction was quenched with 20 cm³ of isopropyl alcohol and the solution was added dropwise with stirring into 2L of methanol. The resulting suspension was filtered and the residue was air dried to yield a white flaky solid (1.5g, 16.5%). Viscosity average molecular mass 1.25 x 10<sup>6</sup>; glass transition temperature (Tg) 148°–155°C; m.p. 165°C. The oligomers were not determined.<sup>4</sup>

## References:

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- 2. Heyns, K.; v. Bebenburg, W. Chem. Ber. 1953, 86, 278.
- 3. *Beilstein* Vol. 12, 3<sup>rd</sup> Suppl. p. 2346.
- 4. The yield of polystyrene was not optimized and oligomers, though present, were not analyzed for mass range and yields.